RESEARCH ARTICLE

Novel diphenyl esters of peptidyl α -aminoalkylphosphonates as inhibitors of chymotrypsin and subtilisin

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Abstract

The activities of novel Cbz-*N*-protected α -aminophosphonic phenyl esters, analogs of leucine (1–15) and phenylalanine (17–29), which are substituted at the phenyl ester rings, as well as of their peptidic derivatives (31–43), were investigated for their inhibitory effects on chymotrypsin and subtilisin. The chemical nature and position of the examined substituents clearly demonstrated a strong structure–activity relationship. Among all synthesized compounds the most potent phosphonic-type inhibitors of subtilisin and chymotrypsin were identified, with k_2/K_1 values 114,380 M⁻¹s⁻¹ and 307,380 M⁻¹s⁻¹, respectively.

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Keywords: Serine protease inhibitors; chymotrypsin inhibitor; subtilisin inhibitor; diphenyl α-aminoalkylphosphonates; peptidyl α-aminoalkylphosphonates

Introduction

Chymotrypsin-like serine proteases such as human mast cell chymase, cathepsin G, and microorganisms' serine proteases of the subtilisin family (S8) are considered to be attractive targets for new inhibitor development and the design of potential new drugs. It was shown that inhibition of the subtilisin-like enzyme of the Cryptosporidium spp. parasite significantly diminished infection in cell culture, emphasizing the potential of this family of enzymes as a drug target¹. A dual-active inhibitor of leukocyte protease cathepsin G and mast cell chymase has been proposed as a drug candidate for the simultaneous treatment of asthma and chronic obstructive pulmonary disease². One of the selective human mast cell chymase inhibitors was active as an anti-inflammatory agent in several animal models of inflammation³. The inhibitors of mast cell chymase may also be useful for preventing cardiovascular disease, and many compounds, including phosphonate-type chymase inhibitor Suc-Val-Pro-Phe^P(OPh), have proven their activity in several animal models⁴.

 α -Aminoalkylphosphonate diphenyl esters and their peptidyl derivatives are potent and selective irreversible inhibitors of different classes of serine proteases⁵⁻⁷. They represent a delicate chemical balance between phosphorus atom electrophilicity and specificity of action. The phosphonate phosphorus atom, in contrast to the routinely used diagnostic agent for the detection of serine protease activity diisopropylphosphofluoridate (DIPF), does not have strong electrophilic properties. Additionally, the peptide fragment of peptidyl α-aminoalkylphosphonate derivatives provides specific interactions with the enzyme, which allows the nucleophilic hydroxyl group of Ser195 located in the active site to attack the moderately electrophilic phosphorus atom of the inhibitor. The stability of the resulting enzyme-phosphonic inhibitor complexes range from 8h for chymotrypsin to more than 3 days for trypsin and human neutrophil elastase7.

Reported α -aminoalkylphosphonic inhibitors of chymotrypsin and chymotrypsin-related enzymes such as cathepsin G and rat mast cell protease II contain unsubstituted

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diphenyl ester rings⁶. Only one publication describes a series of halophenyl diester derivatives of α -aminophosphonic inhibitors of chymotrypsin, which showed higher activity than their unsubstituted counterpart structures⁸.

It is worth mentioning that subtilisin and chymotrypsin are not evolutionarily related⁹. The three-dimensional structures of the molecules are not similar, and the catalytic triad composed of His/Asp/Ser in chymotrypsin occurs in a different sequence in subtilisin, Asp/His/Ser. It is of interest to compare the inhibition of both enzymes by the phosphonate-type inhibitors, and in this report we describe a series of new phosphonic analogs of phenylalanine and leucine as chymotrypsin and subtilisin inhibitors.

Materials and methods

Chemistry

All reagents used in the experiments were purchased from Merck, Sigma-Aldrich, Lancaster, and Fluka. Melting points were determined on a Boëtius, Nagema Rapido, PHMK 05 apparatus. ¹H and ³¹P nuclear magnetic resonance (NMR) spectra were recorded on a Bruker AC-TM DRX 300 (300.13 MHz (¹H), 121.51 MHz (³¹P)) spectrometer.

General procedure for synthesis of Cbz-N-protected α -aminoalkylphosphonate diphenyl esters

N-benzyloxycarbonyl diphenyl of esters α -aminoalkylphosphonates were obtained as racemic mixtures by the α -amidoalkylation reaction of triphenyl phosphite⁶. The triphenyl phosphites were prepared by heating phosphorus trichloride (1 eq) with corresponding phenol (3 eq) at 80°C in acetonitrile for 2 h (Scheme 1). The volatile products were evaporated under reduced pressure and the obtained crude phosphite was used directly in the α -amidoalkylation reaction without purification; to the equimolar amount of phosphite, isovaleric aldehyde, or phenylacetaldehyde, benzyl carbamate and acetic acid were added and the mixture was refluxed for 2-4 h. Reaction progress was monitored by thin layer chromatography (TLC). After the reaction mixture was evaporated, the resulting oil was dissolved in methanol and left for crystallization at -20°C. The obtained product was filtered, washed with cold methanol, and recrystallized if necessary.

General procedure for synthesis of dipeptides with *C*-terminal α-aminoalkylphosphonate diphenyl esters

the dipeptide derivatives For synthesis of of α -aminoalkylphosphonate diphenyl esters, hydrobromides of α -aminoalkylphosphonate diphenyl esters were used (Scheme 2): the racemic mixture of Cbz-N-protected α -aminoalkylphosphonate diphenyl ester was dissolved in 33% HBr/AcOH solution and allowed to react for 2h at room temperature; the reaction mixture was evaporated under reduced pressure, and the resulting oil was dissolved in diethyl ether and left at -20°C for crystallization; the product was then filtered, washed with cold diethyl ether, and dried in air. Boc-L-Pro-OH (1.1 eq), triethylamine (2.5 eq), and HBTU (O-(Benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium hexafluorophosphate) (1.2 eq) were added successively to a solution of the α -aminoalkylphosphonate diphenyl ester hydrobromide (1 eq) in acetonitrile. The reaction was performed at room temperature for 24 h. The solution was diluted with ethyl acetate, washed with brine, 5% KHSO, brine, 5% NaHCO,, and finally with water. The organic layer was dried over MgSO,, filtered, and evaporated. All dipeptides containing C-terminal α-aminoalkylphosphonate diphenyl esters were further purified by column chromatography on silica gel.

General procedure for synthesis of tripeptides containing C-terminal α -aminoalkylphosphonate diphenyl esters

The crude Boc-protected dipeptide containing an α -aminoalkylphosphonate diphenyl ester at the C-terminus was dissolved in a dichloromethane/trifluoroacetic acid (TFA) (1:1, v/v) cleavage mixture, and the progress of the reaction was followed by TLC. When the starting material was not present the mixture was evaporated, re-dissolved in dry toluene, and evaporated under reduced pressure. The resulting oil was dissolved in acetonitrile, and coupling with Boc-L-Val-OH was performed by the method described above (Scheme 2). The final compounds were purified by column chromatography.

Procedure for synthesis of Suc-Val-Pro-Phe^P(OPh),

Compound **42** was used for the synthesis of Suc-Val-Pro-Phe^P(OPh)₂. Boc-Val-Pro-Phe^P(OPh)₂ (1 eq) was dissolved in CH_2Cl_2/TFA (1:1, v/v) solution; the mixture was then evaporated, and after dissolving it in ethyl acetate, succinic anhydride (1 eq) and triethylamine (1.1 eq) were added. After 1 h



Reagents: (a) MeCN, reflux 2 h; (b) AcOH, reflux 2-4 h.

Scheme 1. Synthesis of Cbz-N-protected derivatives of α-aminoalkylphosphonate diphenyl esters. Reagents: (a) MeCN, reflux 2h; (b) AcOH, reflux 2-4h.



Reagents: (a) 33% HBr/AcOH, ~2 h r.t.; (b) Boc-Pro-OH, HBTU, Et₃N, MeCN, 24 h r.t.; (c) CH₂Cl₂:TFA (1:1, v/v), r.t.; (d) Boc-Val-OH, HBTU, Et₃N, MeCN, 24 h r.t.

Scheme 2. Synthesis of peptidyl derivatives of α -aminoalkylphosphonate diphenyl esters. Reagents: (a) 33% HBr/AcOH, ~2h room temperature (r.t.); (b) Boc-Pro-OH, HBTU, Et_xN, MeCN, 24h r.t.; (c) CH₂Cl₂/TFA (1:1, v/v), r.t.; (d) Boc-Val-OH, HBTU, Et_xN, MeCN, 24h r.t.

a second equivalent of triethylamine was added. The reaction was performed at room temperature for 2 h and volatile matter was removed under reduced pressure. The obtained residue was diluted with ethyl acetate, and washed with 5% citric acid solution and water. The organic layer was dried over MgSO₄, filtered, and evaporated, yielding the desired compound.

Compound **1** Yield 38%; m.p. 126°C; ³¹P NMR (CDCl₃): 20.37 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.79 Hz, 6H, 2×CH₃), 1.61–2.16 (m, 3H, CH₂, CH), 2.47 (s, 6H, 2×CH₃), 4.61 (m, 1H, CHP), 5.18 (d, J=10.37 Hz, 1H, NH), 5.06 (m, 2H, ArCH₂O), 6.91–7.30 (m, 13H, Ar-H).

Compound **2** Yield 68%; m.p. 161°C; ³¹P NMR (CDCl₃): 19.03 (s); ¹H NMR (CDCl₃): 0.89 (d, J = 5.87 Hz, 6H, $2 \times CH_3$), 1.19 (s, 18H, $6 \times CH_3$), 1.63–2.09 (m, 3H, CH₂, CH), 4.48 (m, 1H, CHP), 5.05 (m, 2H, ArCH₂O), 6.66–7.27 (m, 13H, Ar-H). *Compound* **3** Yield 22%; m.p. 91–105°C; ³¹P NMR (CDCl₃): 18.18 (s); ¹H NMR (CDCl₃): 0.99 (d, J = 5.52 Hz, 6H, $2 \times CH_3$), 1.73–1.84 (m, 3H, CH₂, CH), 2.00–2.32 (m, $6 \times CH_3$), 4.60 (m, 1H, CHP), 5.08 (m, 2H, ArCH₂O), 6.76–7.34 (m, 9H, Ar-H).

Compound 4 Yield 51%; m.p. 111°C; ³¹P NMR (CDCl₃): 18.75 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.44 Hz, 6H, 2×CH₃), 1.73-1.84 (m, 3H, CH₂, CH), 2.41 (d, J=13.92 Hz, 18H, 6×CH₃), 4.58 (m, 1H, CHP), 5.11 (d, J=10.53 Hz, 1H, NH), 5.03 (m, 2H, ArCH₂O), 6.50-7.37 (m, 9H, Ar-H).

Compound **5** Yield 48%; m.p. 103°C; ³¹P NMR (CDCl₃): 19.03 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.8Hz, 6H, 2×CH₃), 1.22 (m, 12H, 4×CH₃), 1.73–1.82 (m, 3H, CH₂, CH), 2.83 (m, 2H, 2×CH), 4.60 (s, 1H, CHP), 5.15 (s, 1H, NH), 5.17 (m, 1H, ArCH₂O), 6.97–7.35 (m, 13H, Ar-H).

Compound **6** Yield 47%; m.p. 104°C; ³¹P NMR (CDCl₃): 18.84 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.64 Hz, 6H, 2×CH₃), 1.87 (m, 3H, CH₂, CH), 2.29 (m, 12H, 4×CH₃), 4.61 (m, 1H, CHP), 5.13 (m, 2H, ArCH₂O), 6.82–7.33 (m, 11H, Ar-H).

Compound **7** Yield 29%; m.p. 115°C; ³¹P NMR (CDCl₃): 18.40 (s); ¹H NMR (CDCl₃): 1.01 (d, *J*=5.04 Hz, 6H, 2×C**H**₃),

1.91 (m, 3H, CH_2 , CH), 2.25 (m, 12H, $4 \times CH_3$), 4.68 (m, 1H, CHP), 5.20 (m, 2H, ArCH₂O), 6.82–7.43 (m, 11H, Ar-H).

Compound **8** Yield 43%; m.p. 87°C; ³¹P NMR (CDCl₃): 19.53 (s); ¹H NMR (CDCl₃): 0.90 (d, J=5.68 Hz, 6H, 2×CH₃), 1.79 (m, 3H, CH₂, CH), 3.82 (s, 6H, 2×COOCH₃), 4.52 (m, 1H, CHP), 5.01–5.16 (m, 3H, ArCH₂O, NH), 6.92–7.93 (m, 13H, Ar-H).

Compound **9** Yield 38%; m.p. 118°C; ³¹P NMR (CDCl₃): 18.82 (s); ¹H NMR (CDCl₃): (d, J=5.86 Hz, 6H, 2×CH₃), 1.77 (m, 3H, CH₂, CH), 2.31 (d, J=7.10 Hz, 6H, 2×CH₃), 4.61 (m, 1H, CHP), 5.11 (s, 1H, NH), 5.14 (m, 1H, 2H, ArCH₂O), 6.96–7.35 (m, 13H, Ar-H).

Compound **10** Yield 63%; m.p. 107°C; ³¹P NMR (CDCl₃): 18.43 (s); ¹H NMR (CDCl₃): 0.99 (m, 6H, $2 \times CH_3$), 1.81 (m, 3H, CH₂, CH), 2.25 (m, 12H, $4 \times CH_3$), 4.68 (m, 1H, CHP), 4.98–5.16 (m, 3H, ArCH₂O, NH), 6.90–7.30 (m, 11H, Ar-H).

Compound 11 Yield 62%; m.p. 116° C; ³¹P NMR (CDCl₃): 18.64 (s); ¹H NMR (CDCl₃): 1.00 (m, 6H, 2 × CH₃), 1.78 (m, 3H, CH₂, CH), 2.25 (m, 6H, 2 × CH₃), 4.66 (m, 1H, CHP), 4.99–5.16 (m, 3H, ArCH₂O, NH), 6.96–7.35 (m, 13H, Ar-H).

Compound **12** Yield 45%; m.p. 86°C; ³¹P NMR (CDCl₃): 18.99 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.60 Hz, 6H, 2×CH₃), 1.22 (m, 6H, 2×CH₃), 1.72 (m, 3H, CH₂, CH), 2.55 (m, 4H, 2×CH₂), 4.60 (m, 1H, CHP), 5.15 (s, 1H, NH), 5.17 (m, 1H, ArCH₂O), 7.27–7.32 (m, 13H, Ar-H).

Compound **13** Yield 45%; m.p. 142°C; ³¹P NMR (CDCl₃): 19.43 (s); ¹H NMR (CDCl₃): 0.98 (d, J=5.96 Hz, 6H, 2×CH₃), 1.88 (m, 3H, CH₂, CH), 4.63 (m, 1H, CHP), 5.19 (s, 1H, NH), 5.11 (m, 2H, ArCH₂O), 6.90–7.35 (m, 13H, Ar-H).

Compound **14** Yield 61%; m.p. 59–60°C; ³¹P NMR (CDCl₃): 19.29 (s); ¹H NMR (CDCl₃): 0.99 (d, J=5.98 Hz, 6H, 2×CH₃), 1.86 (m, 3H, CH₂, CH), 4.65 (m, 1H, CHP), 4.96–5.18 (m, 3H, ArCH₂O, NH), 6.80–7.36 (m, 13H, Ar-H).

Compound **15** Yield 53%; m.p. 59°C; ³¹P NMR (CDCl₃): 18.76 (s); ¹H NMR (CDCl₃): 0.98 (d, *J*=5.68 Hz, 6H, 2×CH₃), 1.69–2.30 (m, 3H, CH₂, CH), 3.80 (d, *J*=11.17 Hz, 6H, 2×CH₂), 4.65 (m, 1H, CHP), 5.13 (m, 2H, ArCH₂O), 5.19 (s, 1H, NH), 6.40–7.33 (m, 13H, Ar-H).

Compound **16** Yield 37%; m.p. 112°C; ³¹P NMR (CDCl₃): 17.97 (s); ¹H NMR (CDCl₃): 0.98 (d, J = 5.60 Hz, 6H, $2 \times CH_3$), 1.93 (m, 3H, CH₂, CH), 4.65 (m, 1H, CHP), 5.00 (s, 1H, ArCH₂O), 5.11 (d, J = 8.99 Hz, 1H, NH), 7.13–7.32 (m, 15H, Ar-H).

Compound **17** Yield 38%; m.p. 137°C; ³¹P NMR (CDCl₃): 19.51 (s); ¹H NMR (CDCl₃): 3.09 (d, J = 12.32 Hz, 6H, $2 \times CH_3$), 3.4 (m, 2H, CH₂), 4.8 (m, 1H, CHP), 4.95 (s, 2H, ArCH₂O), 5.05 (d, J = 9.96 Hz, 1H, NH), 7.19–7.80 (m, 18H, Ar-H).

Compound **18** Yield 47%; m.p. 112°C; ³¹P NMR (CDCl₃): 17.58 (s); ¹H NMR (CDCl₃): 1.18 (m, 6H, $2 \times CH_3$), 2.52 (m, 4H, $2 \times CH_2$), 3.30 (m, 2H, CH₂), 4.7 (m, 1H, CHP), 4.92 (s, 2H, ArCH₂O), 5.14 (d, *J*=10.2Hz, 1H, NH), 6.91–7.23 (m, 18H, Ar-H).

Compound **19** Yield 38%; m.p. 119°C; ³¹P NMR (CDCl₃): 17.28 (s); ¹H NMR (CDCl₃): 3.16 (m, 6H, $2 \times CH_3$), 3.85 (m, 2H, CH_2), 5.70 (m, 1H, CHP), 5.8 (m, 2H, $ArCH_2O$), 6.07 (d, J=10.43 Hz, 1H, NH), 7.8–8.14 (m, 18H, Ar-H).

Compound **20** Yield 88%; m.p. 99–102°C; ³¹P NMR (CDCl₃): 18.11 (s); ¹H NMR (CDCl₃): 3.40 (m, 2H, C**H**₂,), 3.69 (d, J=8.24 Hz, 6H, 2vC**H**₃), 4.70 (m, 1H, C**H**P), 4.92 (s, 2H, ArC**H**₂O), 5.12 (d, J=10.57 Hz, 1H, N**H**), 6.65–7.23 (m, 18H, Ar-**H**).

Compound **21** Yield 28%; m.p. 129°C; ³¹P NMR (CDCl₃): 17.63 (s); ¹H NMR (CDCl₃): 1.16 (t, J=8.12 Hz, 12H, 4×CH₃), 2.94 (m, 2H, 2×CH), 3.37 (m, 2H, CH₂), 4.77 (m, 1H, CHP), 4.81 (s, 2H, ArCH₂O), 5.36 (d, J=9.95 Hz, 1H, NH), 6.93–7.25 (m, 18H, Ar-H).

Compound **22** Yield 54%; m.p. 99°C; ³¹P NMR (CDCl₃): 18.04 (s); ¹H NMR (CDCl₃): 2.90–3.00 (m, 2H, CH₂), 4.73 (m, 1H, CHP), 4.92 (m, 2H, ArCH₂O), 5.09 (d, *J*=10.50 Hz, 1H, NH), 7.00–7.23 (m, 18H, Ar-H).

Compound **23** Yield 40%; m.p. 103°C; ³¹P NMR (CDCl₃): 17.08 (s); ¹H NMR (CDCl₃): 2.17 (m, 12H, $4 \times CH_3$), 2.9–3.4 (m, 2H, CH₂), 4.8 (m, 1H, CHP), 4.90 (m, 2H, ArCH₂O), 5.15 (d, *J*=10.42 Hz, 1H, NH), 6.86–7.28 (m, 16H, Ar-H).

Compound **24** Yield 22%; m.p. 139°C; ³¹P NMR (CDCl₃): 17.64 (s); ¹H NMR (CDCl₃): 1.09 (m, 18H, $6 \times CH_3$), 3.18 (m, 2H, CH_2), 4.78 (m, 1H, CHP), 4.9 (s, 2H, $ArCH_2O$), 5.02 (d, J=10.0, 1H, NH), 6.71–7.13 (m, 18H, Ar-H).

Compound **25** Yield 10%; m.p. 151–152.5°C; ³¹P NMR $(CDCl_3)$: 17.64 (s); ¹H NMR $(CDCl_3)$: 2.98–3.42 (m, 2H, CH₂), 3.87 (s, 6H, 2×CH₃), 4.83 (m, 1H, CHP), 4.97 (s, 2H, ArCH₂O), 5.12 (d, *J*=10.5 Hz, 1H, NH), 6.7–7.13 (m, 18H, Ar-H).

Compound **26** Yield 40%; m.p. 97–99°C; ³¹P NMR (CDCl₃): 18.98 (s); ¹H NMR (CDCl₃): 2.40 (d, J=6.52 Hz, 6H, 2×CH₃), 2.90–3.36 (m, 2H, CH₂), 4.70 (m, 1H, CHP), 4.93 (s, 2H, ArCH₂O), 5.07 (d, J=10.9 Hz, 1H, NH), 6.69–7.28 (m, 18H, Ar-H).

Compound **27** Yield 41%; m.p. 72°C; ³¹P NMR (CDCl₃): 17.45 (s); ¹H NMR (CDCl₃): 2.2 (m, 12H, $4 \times CH_3$), 2.9–3.3 (m, 2H, CH_2), 4.91 (m, 1H, CHP), 5.30 (d, J=10.42 Hz, 1H, NH), 5.02 (m, 2H, ArCH₂O), 6.54–7.23 (m, 16H, Ar-H).

Compound **28** Yield 50%; m.p. 101–104°C; ³¹P NMR $(CDCl_3)$: 17.91 (s); ¹H NMR $(CDCl_3)$: 2.90–3.02 (m, 2H, CH₂), 4.75 (m, 1H, CHP), 4.91 (m, 2H, ArCH₂O), 5.31 (d, *J*=10.30 Hz, 1H, NH), 7.00–7.23 (m, 18H, Ar-H).

Compound **29** Yield 36%; m.p. 155°C; ³¹P NMR (CDCl₃): 17.36 (s); ¹H NMR (CDCl₃): 2.19 (m, 18H, $6 \times CH_3$), 2.90–3.38 (m, 2H, CH₂), 4.72 (m, 1H, CHP), 4.88(m, 2H, ArCH₂O), 5.1 (d, *J*=10.08 Hz, 1H, NH), 6.66–7.23 (m, 14H, Ar-H).

Compound **30** Yield 48%; m.p. 124.5–125.5°C; ³¹P NMR (CDCl₃): 18.46 (s); ¹H NMR (CDCl₃): 2.98–3.48 (m, 2H, CH₂), 4.82 (m, 1H, CHP), 4.88 (s, 2H, ArCH₂O), 5.15 (d, *J*=10.17 Hz, 1H, NH), 7.06–7.29 (m, 20H, Ar-H).

Compound **31** Yield 43%; m.p. 54°C; ³¹P NMR (CDCl₃): 19.62 (s); ¹H NMR (CDCl₃): 0.81 (m, 6H, $2 \times CH_3$), 1.31 (s, 9H, $3 \times CH_3$), 1.66–2.21 (m, 7H, $3 \times CH_2$, CH), 3.26 (m, 2H, CH₂N), 3.76 (s, 6H, $2 \times CH_3$), 4.17 (m, 1H, CHCO), 4.78 (m, 1H, CHP), 7.27–7.75 (m, 8H, Ar-H).

Compound **32** Yield 66%; m.p. $53-55^{\circ}$ C; ³¹P NMR (CDCl₃): 18.76 (s), 18.58 (s); ¹H NMR (CDCl₃): 0.88 (m, 6H, 2×CH₃), 1.34 (s, 9H, 3×CH₃), 1.62–1.98 (m, 7H, 3×CH₂, CH), 2.53 (m, 18H, 6×CH₃), 3.4 (m, 2H, CH₂N), 4.2 (m, 1H, CHCO), 4.8 (m, 1H, CHP), 6.67–7.31 (m, 4H, Ar-H).

Compound **33** Yield 18%; m.p. 53°C; ³¹P NMR (CDCl₃): 18.78 (s), 18.61 (s); ¹H NMR (CDCl₃): 0.88 (m, 6H, $2 \times CH_3$), 1.36 (s, 9H, $3 \times CH_3$), 1.62–2.02 (m, 5H, $2 \times CH_2$, CH), 2.07–2.09 (m, 6H, $2 \times CH_3$), 2.12–2.21 (m, 2H, CH₂), 3.27 (m, 2H, CH₂N), 4.14 (m, 1H, CHCO), 4.86 (m, 1H, CHP), 6.97–7.28 (m, 4H, Ar-H).

Compound **34** Yield 30%; m.p. 49–51°C; ³¹P NMR (CDCl₃): 18.86 (s); ¹H NMR (CDCl₃): 0.86 (d, J=9.07 Hz, 6H, 2×CH₃), 1.37 (s, 9H, 3×CH₃), 1.63–2.02 (m, 5H, 2×CH₂, CH), 2.12 (m, 12H, 4×CH₃), 2.23 (m, 2H, CH₂), 3.32 (m, 2H, CH₂N), 4.20 (m, 1H, CHCO), 4.76 (m, 1H, CHP), 6.78–7.19 (m, 6H, Ar-H).

Compound **35** Yield 65%; m.p. $39.5-40.5^{\circ}$ C; ³¹P NMR (CDCl₃): 18.78 (s), 18.79 (s); ¹H NMR (CDCl₃): 0.93 (m, 6H, 2×CH₃), 1.44 (s, 9H, 3×CH₃), 1.69–2.04 (m, 7H, 3×CH₂, CH), 2.15–2.30 (m, 6H, 2×CH₃), 3.40 (m, 2H, CH₂N), 4.24 (d, *J*=16.22 Hz, 1H, CHCO), 4.85 (m, 1H, CHP), 7.00–7.28 (m, 8H, Ar-H).

Compound **36** Yield 90%; m.p. 96°C; ³¹P NMR (CDCl₃): 19.00 (s), 18.83 (s); ¹H NMR (CDCl₃): 0.85 (m, 6H, $2 \times CH_3$), 1.37 (s, 9H, $3 \times CH_3$), 1.65–1.90 (m, 7H, $3 \times CH_2$, CH), 3.50 (m, 2H, CH₂N), 3.6 (t, *J*=2.32 Hz, 6H, $2 \times CH_3$), 4.30 (m, 1H, CHCO), 5.2 (m, 1H, CHP), 6.72–7.19 (m, 8H, Ar-H).

Compound **37** Yield 19%; m.p. 69°C; ³¹P NMR (CDCl₃): 17.83 (s), 17.63 (s); ¹H NMR (CDCl₃): 2.33 (m, 9H, $3 \times CH_3$), 1.36–1.98 (m, 4H, $2 \times CH_2$), 3.41 (m, 4H, CH_2 N, CH_2), 4.19 (m, 1H, CHCO), 5.18 (m, 1H, CHP), 7.04–7.67 (m, 15H, Ar-H).

Compound **38** Yield 90%; m.p. 49–52°C; ³¹P NMR (CDCl₃): 18.24 (s), 18.16 (s); ¹H NMR (CDCl₃): 2.5 (s, 9H, $3 \times CH_3$), 1.65–1.90 (m, 4H, $2 \times CH_2$), 2.35 (s, 6H, $2 \times CH_3$), 3.40 (m, 4H, CH_2 N, CH_2), 4.12 (m, 1H, CHCO), 5.00 (m, 1H, CHP), 5.19 (s, 1H, NH), 6.94–7.18 (m, 13H, Ar-H).

Compound **39** Yield 43%; m.p. 59° C;³¹P NMR (CDCl₃): 18.51 (s), 18.63 (s); ¹H NMR (CDCl₃): 0.84 (m, 12H, 4×CH₃), 1.34 (s, 9H, 3×CH₃), 1.65–1.86 (m, 8H, 3×CH₂, 2×CH), 2.13– 2.16 (m, 18H, 6×CH₃), 3.50 (m, 3H, CH₂N), 4.20 (m, 1H, CH), 4.36 (m, 1H, CHCO), 4.7 (m, 1H, CHP), 5.2 (d, *J*=10.52 Hz, 1H, NH), 6.70-7.0 (m, 4H, Ar-H).

Compound **40** Yield 23%; m.p. 62°C; ³¹P NMR (CDCl₃): 19.83 (s), 19.55(s); ¹H NMR (CDCl₃): 0.92 (m, 12H, 4×C**H**₂),

		Cbz-Leu ^p -(O	Ar) ₂				
Subtilisin Chymotry							
Compound	fAr	<i>K</i> _i (μM)	$k_2^2/K_i^2(M^{-1}s^{-1})$	$K_{i}(\mu M)$	$k_2/K_1 (M^{-1}s^{-1})$		
1	4-Mercaptomethylphenyl	>20	<50	0.60 ± 0.096	18300 ± 82		
2	4- <i>t</i> -Butylphenyl	7.92 ± 0.6	253 ± 27	>20	<50		
3	2,3,5-Trimethylphenyl	>20	<50	6.7 ± 0.47	213 ± 70		
4	3,4,5-Trimethylphenyl	>20	<50	0.22 ± 0.027	4425 ± 575		
5	4- <i>iso</i> -Propylphenyl	>20	<50	>20	<50		
6	4,5-Dimethylphenyl	>20	<50	7.6 ± 1.52	306 ± 3		
7	2,5-Dimethylphenyl	>20	<50	>20	<50		
8	4-carboxymethylphenyl	>20	<50	0.16 ± 0.02	$61,770 \pm 5860$		
9	4-Methylphenyl	>20	<50	1.3 ± 0.4	914 ± 228		
10	2,3-Dimethylphenyl	>20	<50	2.2 ± 0.28	3500 ± 455		
11	2-Methylphenyl	>20	<50	10 ± 3	293 ± 102		
12	4-Ethylphenyl	>20	<50	7.6 ± 0.76	173 ± 21		
13	4-Chlorophenyl	>20	<50	1.6 ± 0.4	2170 ± 390		
14	3-Chlorophenyl	>20	<50	2.6 ± 0.13	4430 ± 310		
15	3-Methoxyphenyl	>20	<50	1.0 ± 0.3	8760 ± 1570		
16	Phenyl	>20	<50	2.5 ± 0.5	650 ± 32		

Table 1. K_i , k_2/K_i for inactivation of chymotrypsin and subtilisin by leucine related phosphonates.

Table 2. K_i , k_p/K_i for inactivation of chymotrypsin and subtilisin by phenylalanine related phosphonates.

Cbz-Phe ^p -(OAr) ₂						
		Sub	tilisin	Chymo	trypsin	
Compound	Ar	<i>K</i> _i (μM)	$k_2/K_i (M^{-1}s^{-1})$	<i>K</i> _i (μM)	k_2/K_1 (M ⁻¹ s ⁻¹)	
17	4-Methylsulfonylphenyl	0.076 ± 0.019	47,100±2305	0.016 ± 0.002	307,380±21,500	
18	4-Ethylphenyl	>20	<50	0.044 ± 0.0044	$66,140 \pm 7270$	
19	2-Methylphenyl	>20	<50	>20	<50	
20	4-Methoxyphenyl	>20	<50	0.045 ± 0.0067	$76,000 \pm 9880$	
21	4-Iso-propylphenyl	>20	<50	3.3 ± 0.74	1100 ± 121	
22	4-Chlorophenyl	>20	<50	1.5 ± 0.42	$10,140 \pm 152$	
23	2,3-Dimethylphenyl	>20	<50	0.2 ± 0.04	9500 ± 1330	
24	4- <i>t</i> -Butylphenyl	>20	<50	1.3 ± 0.25	1460 ± 263	
25	4-Carboxymethylphenyl	2.74 ± 0.44	730 ± 182	0.01 ± 0.002	$300,000 \pm 60,000$	
26	4-Mercaptomethylphenyl	2.13 ± 0.63	802 ± 144	0.65 ± 0.06	86,000±17,200	
27	4,5-Dimethylphenyl	>20	<50	6.2 ± 0.8	218 ± 5	
28	3-Chlorophenyl	>20	<50	0.14 ± 0.037	$46,600 \pm 12,120$	
29	3,4,5-Trimethylphenyl	>20	<50	15 ± 2.2	94 ± 1	
30	Phenyl	6.37 ± 1.3	760 ± 228	4.7 ± 0.8	3460 ± 242	

Table 3. K_{i} , k_{2}/K_{i} for inactivation of chymotrypsin and subtilisin by dipeptidyl derivatives of leucine related phosphonates.

	Boc-Pro-Leu [®] (OAr) ₂							
		Subtilisin		Chymotrypsin				
Compound	Ar	$K_{i}(\mu M)$	$k_2/K_1 (M^{-1}s^{-1})$	<i>K</i> _i (μM)	k_2/K_1 (M ⁻¹ s ⁻¹)			
31	3-Carboxymethylphenyl	1.35 ± 0.054	$10,420 \pm 1250$	2.7 ± 0.5	5330 ± 37			
32	3,4,5-Trimethylphenyl	>20	<50	4.6 ± 0.6	1780 ± 213			
33	2-Methylphenyl	>20	<50	0.5 ± 0.015	3215 ± 96			
34	3,4-Dimethylphenyl	36 ± 0.11	62 ± 4.4	10.8 ± 2.2	53 ± 8.5			
35	4-Methylphenyl	2.28 ± 0.091	1100 ± 220	>20	<50			
36	4-Methoxyphenyl	7.05 ± 0.63	1930 ± 251	>20	<50			

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		Boc-Pro-Ph	e ^P (OAr) ₂		
		Subtilisin		Chymotrypsin	
Compound	Ar	$K_{i}(\mu M)$	$k_2^{}/K_1^{}(\mathrm{M}^{-1}\mathrm{s}^{-1})$	$K_{i}(\mu M)$	$k_2/K_1 (M^{-1}s^{-1})$
37	Phenyl	12.1±3	386 ± 77	10.4 ± 2.4	170 ± 30
38	4-Mercaptomethylphenyl	5.3 ± 0.21	1560 ± 141	0.73 ± 0.073	1600 ± 160

	Table 4.	$K_{a}, k_{a}/K_{a}$ for inactivation	on of chymotrypsin and	subtilisin by dipeptidyl deriva	tives of phenylalanine	related phosphonates
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Table 5. K_{i} , k_2/K_1 for inactivation of chymotrypsin and subtilisin by tripeptidyl derivatives of leucine related phosphonates.

Boc-Val-Pro-Leu $^{P}(OAr)_{2}$						
		Subtilisin		Chymotrypsin		
Compound	Ar	$K_{i}(\mu M)$	$k_2^2/K_1^2(M^{-1}s^{-1})$	<i>K</i> _i (μM)	$k_2^{}/K_{ m i}^{}({ m M}^{-1}{ m s}^{-1})$	
39	3,4,5-Trimethylphenyl	>20	<50	0.33 ± 0.0033	9230 ± 1020	
40	4-Methoxyphenyl	0.06 ± 0.004	$13,760 \pm 0$	1.79 ± 0.4	8100 ± 73	
41	4- <i>t</i> -Butylphenyl	>20	<50	0.75 ± 0.067	831±78	

Table 6. $K_i, k_s/K_i$ for inactivation of chymotrypsin and subtilisin by tripeptidyl derivatives of phenylalanine related phosphonates.

$Boc-Val-Pro-Phe^{P}(OAr)_{2}$						
		Subtilisin		Chymotrypsin		
Compound	Ar	<i>K</i> _i (μM)	$k_2^{}/K_1^{}(\mathrm{M}^{-1}\mathrm{s}^{-1})$	<i>K</i> _i (μM)	$k_2/K_i (M^{-1}s^{-1})$	
42	Phenyl	0.23 ± 0.1	71,180±14,230	0.36 ± 0.054	9750 ± 3900	
43	4-Mercapto- methylphenyl	0.093 ± 0.0065	114,380±4570	0.05 ± 0.02	77,600±7760	
Suc-Val-Pro-Phe ^P (OPh) ₂	_	2.06 ± 0.16	3430 ± 343	0.074 ± 0.019	118,250±23,650	

1.36 (s, 9H, $3 \times CH_3$), 1.60–2.10 (m, 8H, $3 \times CH2$, $2 \times CH$), 3.62– 3.68 (m, 2H, CH_2N), 3.69 (s, 6H, $2 \times CH_3$), 4.22 (m, 1H, CH), 4.44 (m, 1H, CHCO), 4.59 (m, 1H, CHP), 5.28 (d, *J*=9.28 Hz, 1H, NH), 6.70–7.19 (m, 4H, Ar-H).

Compound **41** Yield 32%; m.p. 52–58°C; ³¹P NMR (CDCl₃): 18.69 (s), 18.92 (s); ¹H NMR (CDCl₃): 0.92 (m, 12H, $4 \times CH_3$), 1.22 (m, 18H, $6 \times CH_3$), 1.36 (s, 9H, $3 \times CH_3$), 1.67–1.88 (m, 8H, $3 \times CH_2$, $2 \times CH$), 3.50–3.67 (m, 2H, CH_2 N), 4.12 (m, 1H, CH), 4.60 (m, 1H, CHCO), 4.75 (m, 1H, CHP), 5.10 (d, *J*=8.9Hz, 1H, NH), 5.24 (d, *J*=11.32 Hz, 1H, NH), 6.98–7.24 (m, 4H, Ar-H).

Compound **42** Yield 23%; m.p. 68°C; ³¹P NMR (CDCl₃): 17.35 (s), 17.37 (s); ¹H NMR (CDCl₃): 0.68 (m, 6H, $2 \times CH_3$), 1.20 (s, 9H, $3 \times CH_3$), 1.50–1.79 (m, 5H, $2 \times CH_2$, CH), 3.14–3.39 (m, 2H, CH₂N), 3.4 (m, 2H, CH₂), 3.99 (m, 1H, CH), 4.3 (m, 1H, CHCO), 5.12 (m, 1H, CHP), 5.16 (m, 1H, NH), 6.87–7.07 (m. 3H, Ar-H).

Compound **43** Yield 80%; m.p. 72–76°C; ³¹P NMR (CDCl₃): 18.30 (s), 18.26 (s); ¹H NMR (CDCl₃): 0.88 (m, 6H, $2 \times CH_3$), 1.37 (s, 9H, $3 \times CH_3$), 1.65–1.80 (m, 5H, $2 \times CH_2$, CH), 2.35 (m, 6H, $2 \times CH_3$), 3.15 (m, 2H, CH₂N), 3.5 (m, 2H, CH₂), 4.99 (m, 1H, NH), 4.16 (m, 1H, CHCO), 5.11 (m, 1H, CHP), 5.16 (m, 1H, NH), 6.79–7.01 (m, 8H, Ar-H).

Suc-Val-Pro-Phe^p(OPh)₂ Yield 70%; ³¹P NMR (CDCl₃): 18.48 (s); ¹H NMR (CDCl₃): 1.80 (m, 6H, $2 \times CH_3$), 1.83–1.96 (m, 5H, $2 \times CH_2$, CH), 2.54 (m, 4H, $2 \times CH_2$), 3.20 (m, 2H, CH₂N), 3.5 (m, 2H, CHCO), 4.40 (m, 1H, CH), 5.11 (m, 1H, CHP), 7.07–7.15 (m, 17H, Ar-H, $2 \times NH$), 7.79 (s, 1H, OH).

Enzyme inhibition assay

Bovine chymotrypsin and subtilisin A from Bacillus licheniformis were purchased from Calbiochem and Sigma-Aldrich, respectively. The substrate used for both enzymes, Suc-Ala-Ala-Pro-Phe-AMC, was purchased from Calbiochem. The assay buffers were as follows: 100 mM HEPES, 500 mM NaCl, pH 7.5 containing 9% DMSO (dimethyl sulfoxide) for chymotrypsin; 50 mM Tris, 1M NaCl, pH 7.5 containing 0.01% Triton X-100 for subtilisin. The intensity of the fluorescence was measured using a Molecular Devices Gemini XPS Microplate Spectrofluorometer (ex 350 nm, em 460 nm). Measured $K_{\rm m}$ (Michaelis constant) values were 60 µM (for subtilisin) and 70 µM (for chymotrypsin). The K_i (inhibition constant) values were calculated from the inhibition of AMC (7-amido-4-methyl coumarin) formation by different concentrations of the tested compound. The hyperbolic model of inhibition was used to calculate K_1 and k_2 values¹⁰⁻¹². The inhibitory activity of synthesized compounds was determined by the progress curve method under pseudo-first-order conditions $([I]_0 >> [E]_0,$ and with less than 5% substrate conversion). Control curves in the absence of inhibitor were linear. The rate of substrate hydrolysis was continuously monitored by measuring the rate of increase of fluorescence at 460 nm (λ_{exc} 350 nm). The pseudo-first-order rate constants (k_{obs}) for the inhibition of chymotrypsin and subtilisin as a function of time were determined according to the equation: $AMC_t = v_0 [1 - exp(- (k_{obs}t)]/(k_{obs} + AMC_0)$, where AMC, is the fluorescence intensity at time t_{1} AMC₀ is the fluorescence intensity at time zero, and

 v_0 is the reaction velocity at time zero. By fitting the AMC vs. *t* data to this equation using nonlinear regression analysis the k_{obs} values were obtained. The second-order rate constant (k_2/K_i) was calculated from the slope of the linear part of the plot using the following equation: $k_{obs}/[I] = k_2/(K_i + [I])$. The obtained K_i and k_2/K_i values of chymotrypsin and subtilisin inactivation by phosphonic analogs of leucine and phenylalanine and their di- and tripeptide derivatives are summarized in Tables 1–6.

Results and discussion

Twenty eight novel, substituted at diphenyl ester rings Cbz-*N*-protected α -aminophosphonic analogs of leucine (1–15) and phenylalanine (17–29) as well as their peptidic derivatives (31–43) were synthesized by a previously described method⁶. Cbz-*N*-protected α -aminophosphonic analogs of leucine (16) and phenylalanine (30) unsubstituted at their diphenyl ester rings were included as reference compounds. The synthetic approach is presented in Scheme 1 and starts with the synthesis of triphenyl phosphites.

An α -amidoalkylation reaction of triphenyl phosphites gives the desired Cbz-*N*-protected α -aminophosphonates as the racemic mixture, with a yield ranging from 10 to 68%. For the synthesis of their peptidyl derivatives, a standard HBTU coupling method was applied using hydrobromide salts of parent Cbz-deprotected α -aminoalkylphosphonate diphenyl esters (Scheme 2).

Inhibition of serine proteases by diphenyl α -aminophosphonate derivatives is a two step process (Scheme 3). The first step leads to the formation of a reversible noncovalent complex (K_i). The next step involves a first order reaction resulting in the formation of a covalent phosphonylated enzyme (k_i).

The overall inactivation rate constant (k_2/K_i) was measured by the progress curve method. We have set up an arbitrary activity criteria utilizing inhibitors with $K_i > 20$ μ M and $k_2/K_i < 50 \, \text{M}^{-1}\text{s}^{-1}$ as inactive compounds. Inactivation rate constants obtained for subtilisin and chymotrypsin by Cbz-Leu^P(OAr)₂ are shown in Table 1. Among all Cbz-Leu^P(OAr)₂ derivatives tested against subtilisin, only the structure bearing 4-*t*-butylphenyl diesters (**2**) showed moderate activity $(k_2/K_i=253 \, \text{M}^{-1}\text{s}^{-1})$. In contrast, chymotrypsin was inhibited by several phosphonic leucine analogs with compound **8**, a 4-carboxymethylphenyl ester derivative displaying the highest inhibition rate constant value of the group at 61,770 $\text{M}^{-1}\text{s}^{-1}$.

The high inhibitory potency of compound **8** is the result of increased electrophilicity of the phosphorus atom

$$E + I \xrightarrow{k_1} E^{\cdots} I \xrightarrow{k_2} EI$$

Scheme 3. Kinetic scheme for serine protease inhibition by diphenyl α -aminophosphonate derivatives.

caused by strong electrowithdrawing properties of the 4-carboxymethyl group. However, we do not consider these types of compounds to have practical value due to their increased sensitivity to hydrolysis. In contrast, derivatives **1** (4-mercaptomethylphenyl esters) and **15** (3-methoxyphenyl esters) do not carry an electrowithdrawing moiety and their high activity is most likely a result of additional interactions within the enzyme leaving the group binding side (S1', S2' pockets). Compounds **1** and **15** are the most potent single α -aminophosphonic leucine analog inhibitors of chymotrypsin reported so far, with k_2/K_i values of 18,300 M⁻¹s⁻¹ and 8760 M⁻¹s⁻¹, respectively.

The activities of various diphenyl esters of phenylalanine analogs are presented in Table 2. For subtilisin the 4-carboxymethylphenyl (25) and 4-mercaptomethylphenyl (26) esters were found to be the most potent inactivators; however, their activity was not better than the unsubstituted diphenyl ester derivative 30. It seems that the substitution at the phenyl diester ring does not improve activity against subtilisin. In other words, the leaving group side of subtilisin is rather small, and does not accommodate the phenyl rings of any of the tested substituents. Interestingly, no such restriction was noticed for leucine analog 2. Derivative 17 with a 4-methylsulfonyl substitution had the highest potency against subtilisin with the k_2/K_1 value 47,100 M⁻¹s⁻¹; however, the presence of a strongly electron-accepting moiety in the para position on the phenyl phosphonate ester makes this derivative hydrolytically unstable and without practical value.

Compounds **17** and **25** are also the most potent inhibitors of chymotrypsin within phosphonic analogs of phenylalanine, with k_2/K_1 values of 307,380 M⁻¹s⁻¹ and 300,000 M⁻¹s⁻¹ respectively. However, for reasons described above they do not possess practical application potential.

Dipeptide derivatives of α -aminoalkylphosphonate diphenyl esters with a Cbz-*N*-protecting group are the only ones reported in the literature. In general, such compounds display poor inhibitory potency against serine proteases compared even to single diphenyl phosphonic amino acid analogs^{6,7}. Surprisingly, the Boc-protected dipeptidyl derivatives show quite good activity, especially against subtilisin (demonstrated in Tables 3 and 4).

For example, Boc-Pro-Leu^P(OAr)₂, derivative **36** with a 4-methoxy substitution at the phenyl ester ring, displayed K_i =7.05 µM and k_2/K_i =1930 M⁻¹s⁻¹ against subtilisin. The 4-mercaptomethyl derivative **38** is a good peptidic inhibitor of subtilisin, with K_i =5.3 µM and k_2/K_i =1560 M⁻¹s⁻¹, and it also inactivates chymotrypsin with K_i =0.73 µM and k_2/K_i =1600 M⁻¹s⁻¹.

The inhibition rate constants measured for tripeptide derivatives **39–43** are presented in Tables 5 and 6. Tripeptides containing leucine-related phosphonates are relatively weak chymotrypsin inhibitors, but derivative **40** ($K_i = 0.06 \mu M$ and $k_2/K_i = 13,760 M^{-1}s^{-1}$) is one of the most active subtilisin inhibitors found in this study. It is worth noting that no activity was found for tripeptide **41** with a 4-*t*-butylphenyl substitution. The corresponding single α -aminophosphonic acid derivative **2** was the only one active toward subtilisin in series **1–15**. It could be speculated that the observed significant difference in activity resides in the mode of compound binding. The relatively outstanding activity of **2** could be the result of binding the 4-*t*-butylphenyl ester moiety in the S1 pocket of the enzyme in addition to "normal binding" (where the 2-methylpropyl constituent is bound to the S1 pocket). In the case of tripeptide derivative **41** with extended binding through the S3–S1 pockets this additional mode of binding is not possible.

The activity of tripeptides with a *C*-terminal phosphonic analog of phenylalanine, **42** and **43**, is similar to that of the most potent known phosphonate chymotrypsin inhibitor Suc-Val-Pro-Phe^P(OPh)₂^{6.7}. However, compound **43** with 4-mercaptomethylphenyl ester rings is the most active sub-tilisin inhibitor found in this study, with $K_i = 0.093 \,\mu\text{M}$ and $k_2/K_i = 114,380 \,\text{M}^{-1}\text{s}^{-1}$.

In conclusion, we have investigated the influence of substitutions at diphenyl ester rings in a series of α -aminoalkylphosphonates and their peptidyl derivatives on the inhibition of subtilisin and chymotrypsin. Our study clearly demonstrated a strong structure-activity relationship, and a new, highly potent phosphonate-type inhibitor of subtilisin, **43**, was identified. For chymotrypsin, it seems, the substitution of the phosphonate diphenyl ester rings by a variety of different substituents at different positions does not significantly improve inhibitory activity.

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